



# **MicroScale Conductivity Temperature Instrument**

Model 125

Operator's Manual

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# MicroScale Conductivity-Temperature Instrument Model 125

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## INTRODUCTION

The Model 125 MicroScale Conductivity-Temperature Instrument (MSCTI) is designed to measure the temperature and electrical conductivity of water solutions containing conductive ions. The MSCTI provides analog voltage outputs that are functions of the solution's electrical conductivity and temperature. The instrument is intended for use in moving solutions where spatial resolution and time response are important.

## DOs AND DON'Ts

Always use fresh, clean solutions. Large contaminants such as algae flakes can become caught on the sensor's electrodes and cause calibration shifts. If possible, position the sensor so that it points directly upstream in the flow. This orientation is less susceptible to contamination.

Debris caught on the electrodes can sometimes be removed by gently rinsing them under a faucet or using a squeeze bottle. Never tap the stainless steel shaft. This causes vibrations that can cause the glass sensor to become cracked.

Avoid solutions that contain oil or slime. The conductivity electrodes are platinized to increase their ability to pass electrical current into the solution. This plating is sponge-like. If the plating becomes fouled with oil or slime, then the electrodes will have difficulty conducting electricity. The 'PHASE ERROR' light on the electronic bridge's front panel comes on when this condition occurs.

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Minimize the time the sensor is exposed to air after calibration. Calibration shifts will occur if the air exposure time is longer than a few seconds. When moving the sensor from one solution to another, take advantage of the fact that a small drop of solution will cling to the sensor's electrodes and protect them.

Always rinse the sensor with fresh water after use.

Never expose the sensor to water unnecessarily.

The preamplifier is watertight and may be submerged. The cable may also be submerged. The connector to the electronic bridge is not watertight. Do not allow water to get into this connector. If this happens, then the instrument will not operate properly and can be permanently damaged. When storing or transporting a wet sensor assembly, be sure that the connector is kept at the highest point. Small drops of water tend to run down along the cable and into the connector if allowed.

The preamplifier cable is a complicated instrumentation cable. It is important that the cable jacket not allow water to enter the cable. Never allow the cable to be damaged. Don't allow it to lay on the floor where it can be stepped on or have equipment rolled over it.

## SPECIFICATIONS

CONDUCTIVITY	
Measurement equation	$V_o = G * C + V_{off}$
8 hour calibration stability	Better than 1% of C reading
Time response	-3db at approx. 800 hz
Spatial response	-3db at approx. 4 cyc/cm
Noise	< 1mV RMS 10 Hz to 1 kHz
Output voltage range	+/- 5 volts
Linear conductivity range	0.5 to 800 mS/cm

TEMPERATURE	
Measurement equation	$V_o = G \exp (A+B/T) + V_{off}$
8 hour calibration stability	Better than 0.01 deg C
Time response	Approx. 7E-3 seconds
Noise	< 1mV RMS 10 Hz to 200 Hz
Output voltage range	+/- 5 volts
Temperature range	-10 to 100 deg C

## PANEL FUNCTIONS

The Model 125 front panel has several displays and controls. Each of these is described below:

**POWER** - This switch controls power to the electronic bridge and preamplifier assembly. The adjacent light is 'on' when power is applied. Note that the switch simply interrupts power from the power supply module and does not control the power supply module in any way.

**GAIN - TEMPERATURE** - This controls 'G' in the temperature measurement equation mentioned above. 'CW' operation of this adjustment increases 'G'.

**GAIN - CONDUCTIVITY** - This controls 'G' in the conductivity measurement equation mentioned above. 'CW' operation of this adjustment increases 'G'.

**PHASE ERROR** - This lamp comes on when the sensor's electrodes have become damaged or fouled and indicates that the conductivity cell is having difficulty passing current into the solution. Momentary phase errors cause this lamp's output to latch for about one second.

**OVER RANGE** - This lamp comes on when either the conductivity or temperature circuits have voltages within them that are outside of the operating range. Having 'G' settings that are too high usually causes this condition. Momentary over ranges cause this lamp's output to latch for about one second.

The Model 125 back panel contains several connectors. The function of each is fairly obvious except for the circuit outputs that are BNCs. The outer BNC supplies Conductivity  $V_o$ . The inner BNC supplies Temperature  $V_o$ .

## GROUNDING

The Model 125 must always be used with grounded solutions. You have the option of grounding the solution yourself or using the ground supplied by the preamplifier-sensor shaft. The preamplifier-sensor shaft is wired through the cable to JMP 1 on the main bridge's circuit board, near the preamp connector. JMP 1 is accessible by removing the two screws that hold the bridge's cover. The Model 125 is shipped with JMP 1 shorted, connecting the preamp-sensor shaft to circuit ground. If you break this connection, then the preamp-sensor shaft and the sensor float free of circuit ground. This allows you to provide ground to the solution in whatever way that is required. The Model 125 circuit ground must be connected to this ground through the outer side of either BNC on the back panel.

## CALIBRATION

Precision Measurement Engineering (PME) performs temperature calibration when the Model 125 is initially supplied. In this condition, a small piece of tape is placed over the GAIN-TEMPERATURE potentiometer access hole. This is intended to prevent accidental adjustment of this potentiometer creating an uncalibrated condition. The conductivity channel is not calibrated by PME. When a new sensor is installed, the customer must perform both calibrations.

Calibration of the Model 125 is performed in three stages: 'G' setup for proper range, calibration data acquisition, and numerical analysis. Each of these is discussed below:

**G SETUP** - It is important that the instrument does not go out of range during an experiment. To insure that it does not, you must first set both 'G' adjustments at the maximum experimental conditions. The procedure is:

1. Turn both gain pots fully CCW. The over range lamp should be dark.
2. Place the sensor in a solution at the lowest temperature expected during the experiment. Adjust the temperature gain pot CW until the over range light comes on. Back the pot off CCW  $\frac{1}{2}$  to 1 turn. If the light does not come on, then leave the pot set fully CW.
3. Place the sensor in a solution at the highest

conductivity expected. Adjust the conductivity gain pot CW until the over range light comes on. Back the pot off CCW ½ to 1 turn. If the light does not come on, then leave the pot set fully CW.

**CALIBRATION DATA ACQUISITION-** This process depends on the accuracy that you need in your experiment. Calibration of the Model 125 depends on laboratory standards for temperature and conductivity that you supply. The Model 125 will not measure more accurately than these standards. The procedure for simple calibration using a 0.1 deg C resolution thermometer and one solution of known conductivity is given below.

1. Turn the electronic bridge off. Disconnect the preamplifier-sensor cable. Turn the electronic bridge on. Record the voltage at the temperature output. This voltage is temperature Voff. It should be approximately -5 volts. Turn the power off and reconnect the cable. Turn the power on.
2. Expose the sensor to two stirred solutions at temperatures approximately bracketing the expected temperature span of the experiment. Record the temperature from the reference thermometer and the voltage at the temperature output.
3. Hold the sensor with the tip up and gently shake the water from the electrodes. Record the voltage at the conductivity output. This voltage is conductivity Voff. It should be approximately -5 volts.
4. Place the sensor in a solution of known conductivity that is near the highest conductivity expected. (Conductivity can be calculated from density and temperature or salt concentration and temperature or otherwise known.) Record the conductivity and the voltage at the conductivity output.

**NUMERICAL ANALYSIS-** A numerical means for connecting voltage outputs from the Model 125 with their corresponding conductivity and temperatures is described below.

1. Use the temperature response equation given in the Specifications section. Include 'G' in the coefficients 'A' and 'B'. Find a simultaneous solution to the equation:

$$\ln(V(T)-V_{off}) = A + B/T \quad (T \text{ in deg K})$$

for 'A' and 'B' by using the two data points. If your reference measurement of 'T' was absolutely certain, then this equation will give fit errors of about 0.05 deg C over a 20 deg C span.

2. Use the conductivity response equation given in the Specifications section. Solve:

$$V(C) - V_{off} = G * C$$

for 'G' using the data point measured above. If your measurement of conductivity was absolutely certain, then this equation will give fit errors of less than 1% of the reading over the linear range of conductivity.

## MAINTENANCE

Maintenance of the Model 125 consists of keeping the sensor, preamplifier, and cable assembly free from physical damage, rinsed off, and dry.

From time to time a sensor assembly will need to be replaced. The procedures are:

1. Remove the outer screw (screw #1) nearest the cable end of the preamplifier. Place it in a cup to prevent loss. It is very small. Refer to Sensor Replacement drawing.
2. Grip the stainless steel shaft and carefully slide the preamplifier cover toward the sensor until all the interior parts are exposed.
3. Clip the six wires (4 - conductivity and 2 - temperature) about 1 cm from each connecting point on the circuit board. Do this so that some of the color from each wire remains connected to the circuit board. This will help when soldering the new wires.
4. Loosen the set-screw (screw #2) in the shaft retaining bracket at the shaft end of the preamplifier circuit board.
5. Withdraw the stainless steel shaft. Slide the preamplifier cover off the stainless steel shaft. Discard the used shaft.
6. Inspect the o-ring at the cable end of the preamplifier for any cuts or debris. If detected, then replace. Next, apply a small amount of silicon oil around the o-ring.
7. Remove the outer screw (screw #3) on the preamplifier cover that holds the nosepiece. Inspect

the o-ring for cuts or debris. If detected, then replace. Apply a small amount of silicon oil around the o-ring.

8. The nosepiece has a smaller, inner o-ring. Carefully remove it and inspect for cuts or debris. If detected, then replace. Apply a small amount of silicon oil around the o-ring before replacing it back into the nosepiece.
9. Slide the nosepiece onto the preamplifier cover. Be sure not to cut the outer o-ring with the cover. Install the outer screw (screw #3). Be careful not to over tighten.
10. Insert the o-ring expander into the stainless steel shaft where the wires protrude. Insert the wires through the clear, plastic tubing. The fit is not tight and the expander will fall to the side. See the Sensor Insertion drawing.
11. Apply a small amount of silicon oil to the end of the new stainless steel shaft and to the o-ring expander. **Do not put silicone oil on the clear, plastic tubing since it will be inserted into the shaft.**
12. Insert the wires through the nosepiece and carefully slide the preamplifier cover onto the o-ring expander. While holding the preamplifier cover, position the o-ring expander so that it is flush with the stainless steel shaft. Keep the expander and shaft in a straight, horizontal position. Gently push the preamplifier cover over the expander/shaft section and onto the shaft being careful not to cut the inner o-ring. See the Shaft Insertion drawing.
13. Once the preamplifier cover is on the shaft, remove the o-ring expander and insert the wires through the shaft-retaining bracket. Insert the shaft fully into the bracket and tighten the set-screw (screw #2).
14. One at time, unsolder the old wire on the circuit board and solder the new wire (that has been stripped) onto the connecting point. Be careful not to switch the colors. If you become confused, then refer to the Sensor Connection drawing. If you coat the stripped end of the wire with solder flux, then it becomes much easier to solder.
15. Be sure to inspect the solders and cut any exposed-stripped wire.
16. Carefully slide the preamplifier cover back into place over the circuit board. To insure that the cover does not cut the wires, be sure that they are



positioned near the center of the circuit board. Be careful not to cut the o-ring near the cable end as the cover slides over it.

17. Install the outer screw (screw #1). Be careful not to over tighten. Wipe off any remaining silicon oil on the shaft or preamplifier cover.

## REFERENCES

Thermistor Calibration: Thermometrics (1995), **Thermistors Catalog**, Thermometrics, Inc., 808 U.S. Hwy 1, Edison, New Jersey 08817, Phone 1(800) 246-7019. This catalog is free upon request from Thermometrics.

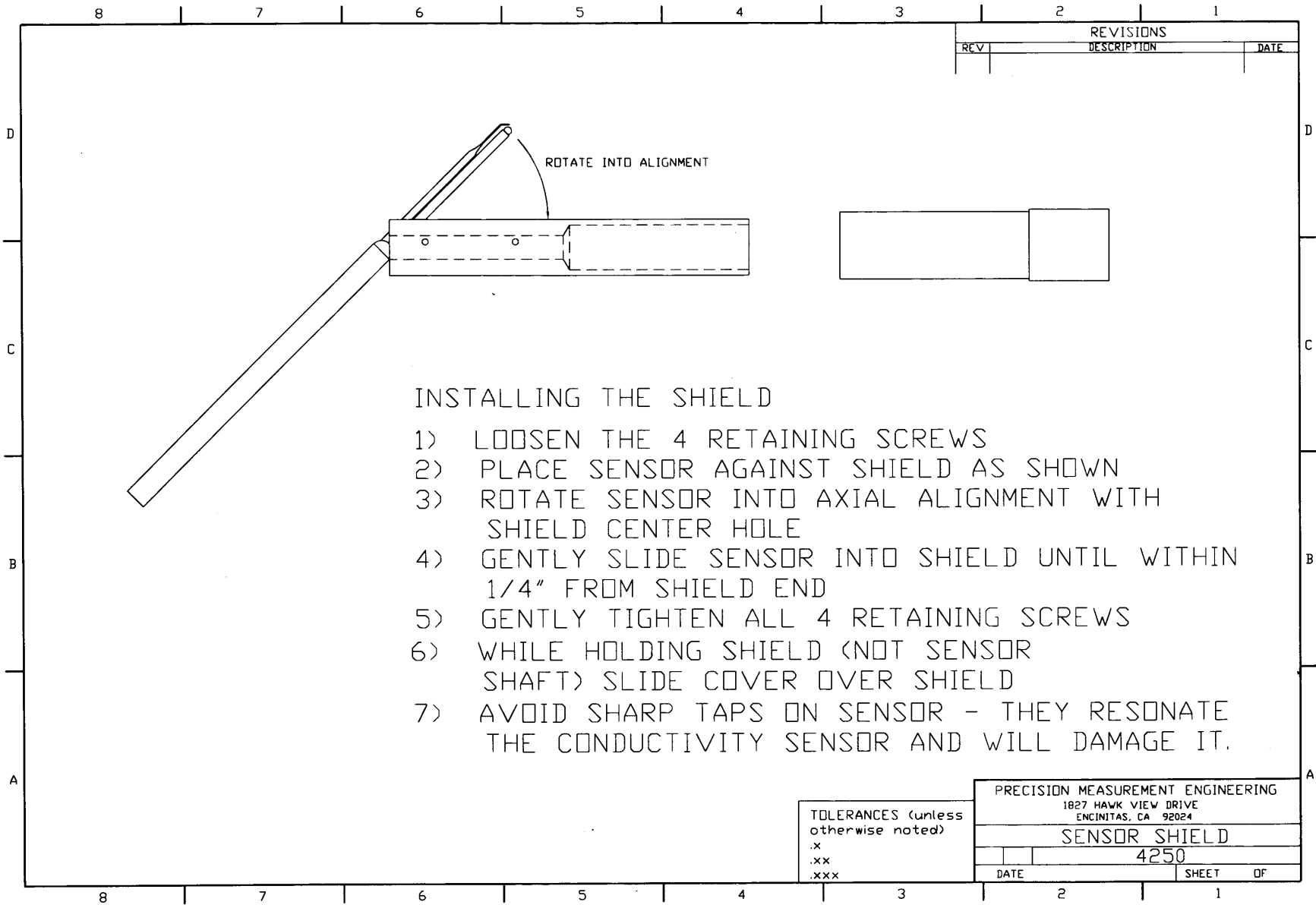
Conductivity for Sodium Chloride Solutions: Chiu, Ying-Check and Fuoss, Raymond M., "Conductance of the Alkali Halides XII. Sodium and Potassium Chloride in Water at 25 C." **Journal of Physical Chemistry**, Vol. 72, # 12, (1968), pp. 4123 - 4129.

Hewitt, G.F., "Tables of the Resistivity of Aqueous Sodium Chloride Solutions", Atomic Energy Research Establishment (Great Britain), **Report R3497**, (1960).

Surdo, Antonio Lo, Alzda, E.M. and Millero, F.J., "The PVT Properties of Concentrated Aqueous Electrolytes. I. Densities and Apparent Molal Volumes of NaCl, Na<sub>2</sub>SO<sub>4</sub>, MgCl<sub>2</sub>, and MgSO<sub>4</sub> solutions from 0.1 to Saturation and from 273.15 to 323.15 K", **Journal of Chemical Thermodynamics**, Vol. 14 (1982), pp. 649 - 662.

Conductivity for Seawater: "Special Issue on the Practical Salinity Scale", **IEEE J. of Oceanic Engineering**, Vol. OE-5, #1 (1980).

MSCTI Instrument: Head, M. J., "The Use of Miniature Four-Electrode Conductivity Probes for High Resolution Measurement of Turbulent Density or Temperature Variations in Salt-Stratified Water Flow", **Ph.D. Thesis, University of California, San Diego**, 1983.



REVISIONS		
REV	DESCRIPTION	DATE

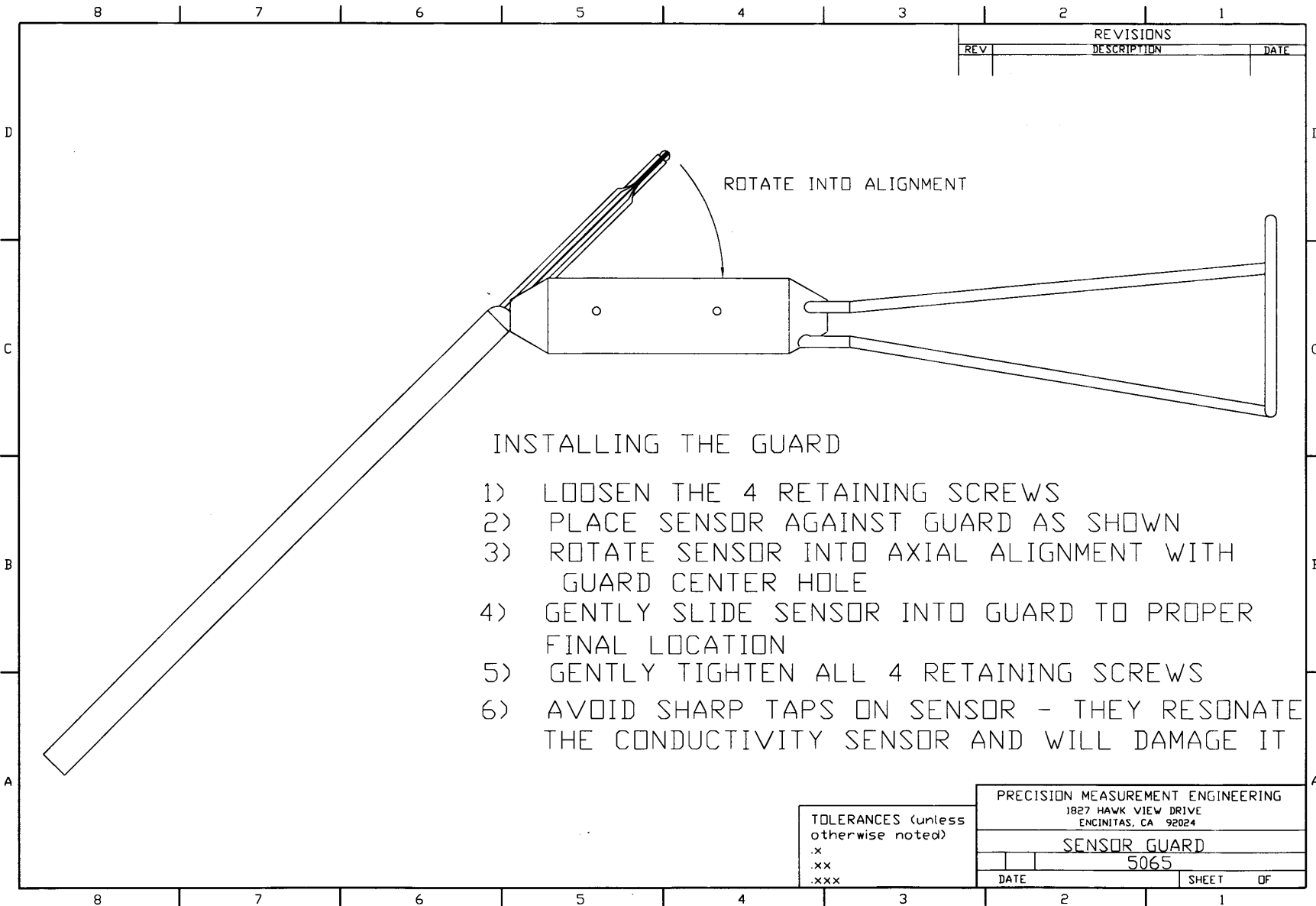
ROTATE INTO ALIGNMENT

INSTALLING THE SHIELD

- 1) LOOSEN THE 4 RETAINING SCREWS
- 2) PLACE SENSOR AGAINST SHIELD AS SHOWN
- 3) ROTATE SENSOR INTO AXIAL ALIGNMENT WITH SHIELD CENTER HOLE
- 4) GENTLY SLIDE SENSOR INTO SHIELD UNTIL WITHIN 1/4" FROM SHIELD END
- 5) GENTLY TIGHTEN ALL 4 RETAINING SCREWS
- 6) WHILE HOLDING SHIELD (NOT SENSOR SHAFT) SLIDE COVER OVER SHIELD
- 7) AVOID SHARP TAPS ON SENSOR - THEY RESONATE THE CONDUCTIVITY SENSOR AND WILL DAMAGE IT.

TOLERANCES (unless otherwise noted)  
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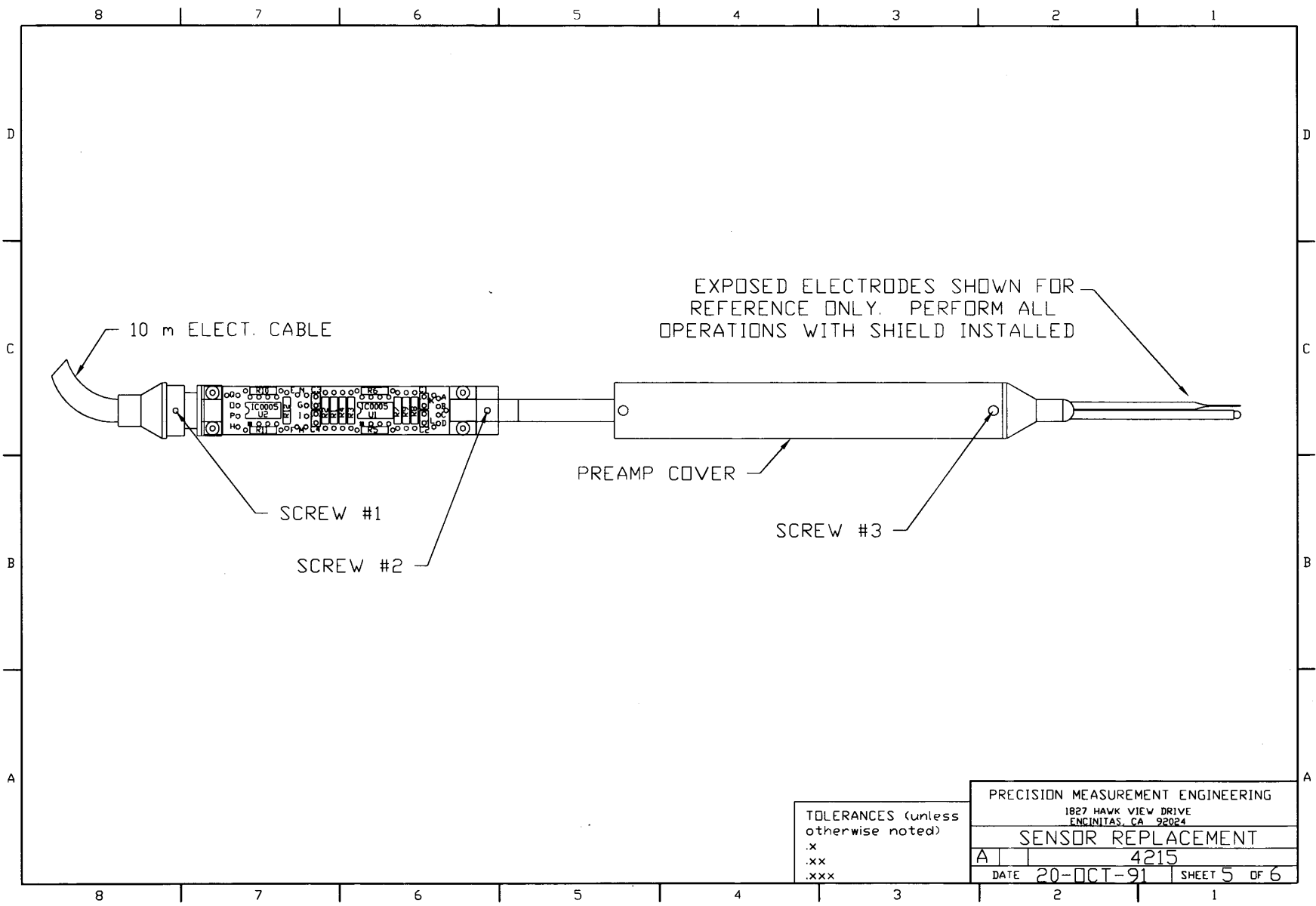
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INSTALLING THE GUARD

- 1) LOOSEN THE 4 RETAINING SCREWS
- 2) PLACE SENSOR AGAINST GUARD AS SHOWN
- 3) ROTATE SENSOR INTO AXIAL ALIGNMENT WITH GUARD CENTER HOLE
- 4) GENTLY SLIDE SENSOR INTO GUARD TO PROPER FINAL LOCATION
- 5) GENTLY TIGHTEN ALL 4 RETAINING SCREWS
- 6) AVOID SHARP TAPS ON SENSOR - THEY RESONATE THE CONDUCTIVITY SENSOR AND WILL DAMAGE IT

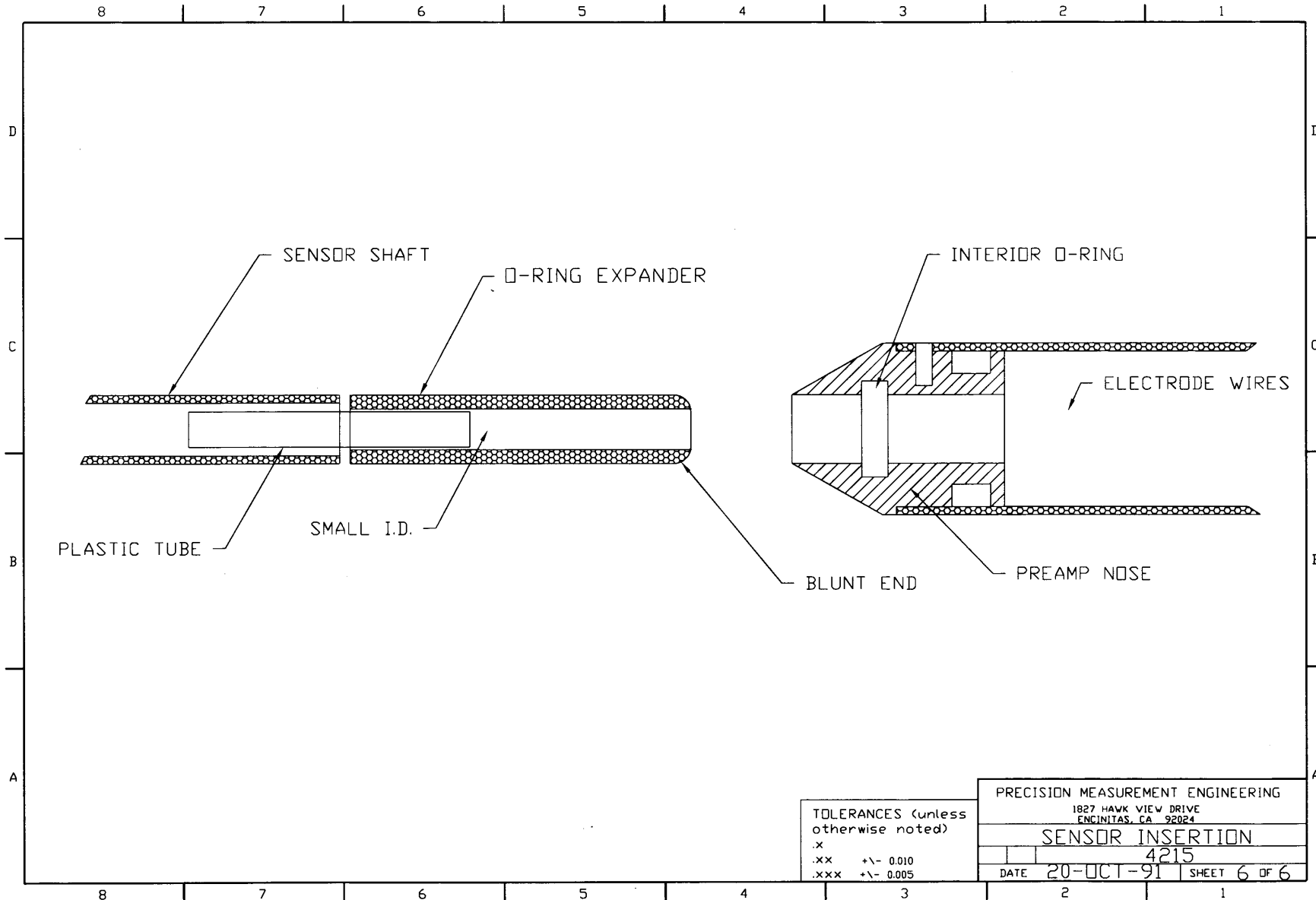
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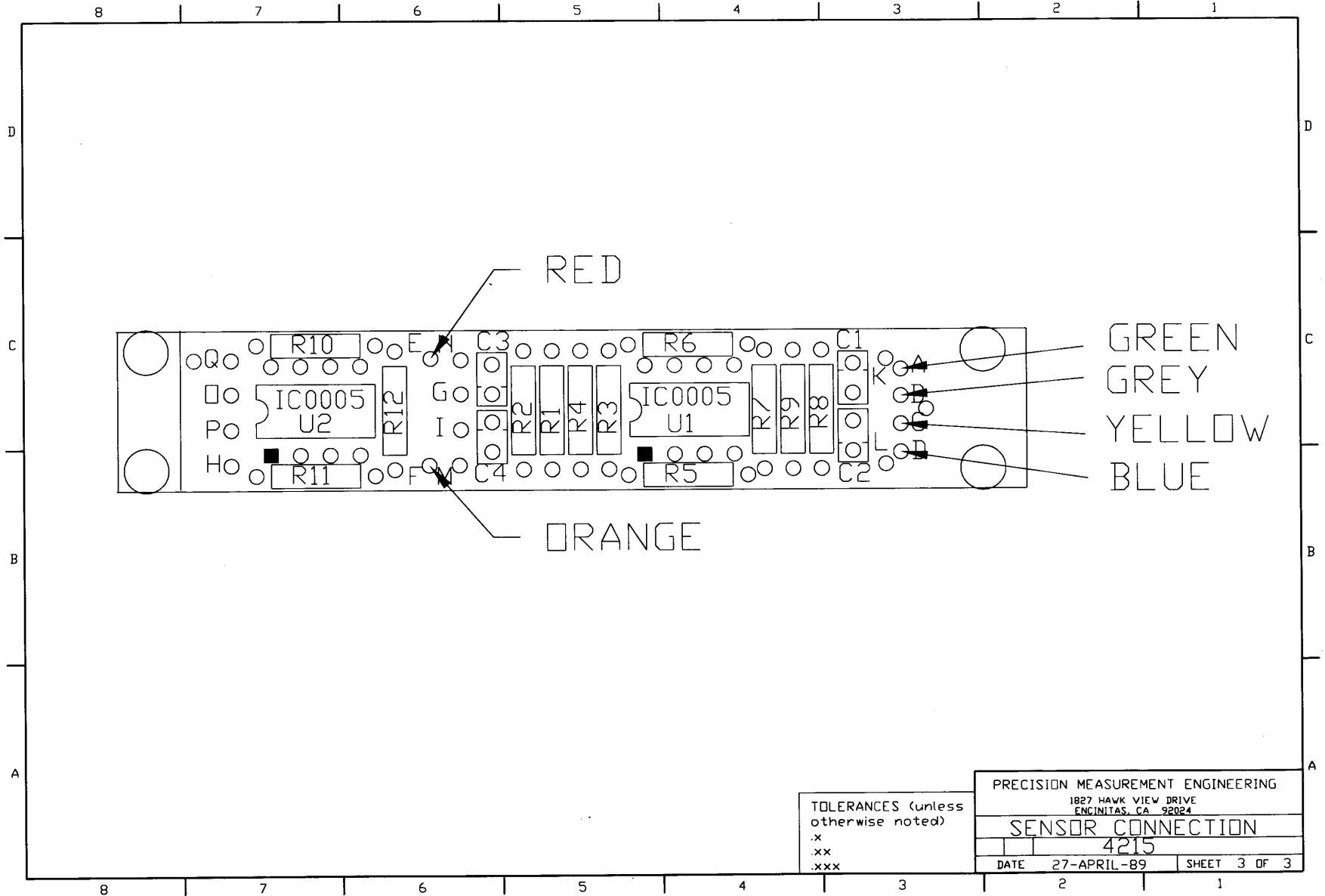
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TABLES OF THE RESISTIVITY OF  
AQUEOUS SODIUM CHLORIDE SOLUTIONS

BY

G. F. HEWITT

ABSTRACT

Tables of the specific resistivity of NaCl solution at 18°C in terms of weight percentage have been derived and are presented. A method of correction for temperature is suggested and tables of temperature coefficient given.

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HARWELL.

October, 1960

HL 60/5450 (S.C. 2)

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### Table

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2.	Temperature coefficients for low-concentration sodium chloride solutions	14.
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1. Introduction

In work previously reported (1) the author made a critical assessment of the data on the resistivity of sodium chloride solutions and suggested methods of using these data for the interpretation of practical measurements. The above work involved the conversion of the data from the usual molar form to the temperature - invariant form of weight percentage. Graphs were prepared of the data at 18°C and temperature coefficient ( $b_t$ ) defined by the equation:

$$R_{18} = R_t [1 + b_t(t - 18)] \quad \sigma_t = \sigma_{18} (1 + b_t(t - 18)) \quad (1)$$

where  $R_{18}$ , the specific resistivity at 18°C, and  $R_t$ , the value at temperature  $t$ °C, were correlated. Sodium chloride solutions have recently been extensively used by the author in studying two-phase gas-liquid flow in pipes. In this work conductometric methods have been used for the determination of film thickness and it was found useful to prepare a tabulation of resistivities and temperature coefficients making use of the authors earlier work. This tabulation was accomplished by writing simple programmes for the Harwell "Mercury" computer. The object of the present report is to present these tables and the equations used in their derivation.

2. Tabulation of data at 18°C

To facilitate interpolation a factor X was defined:

$$X = \% \text{ NaCl} \times \text{Specific resistivity} \quad (2)$$

X varied more slowly with concentration than did specific resistivity. The following polynomial expression were obtained for X:

$p (= \% \text{ NaCl}) < 0.1$

$$0 < 0.01 \quad X = \sum_{n=0}^{n=3} a_n (\sqrt{p})^n \quad (3)$$

where

$$\begin{aligned} a_0 &= 53.5590 \\ a_1 &= 24.2130 \\ a_2 &= -138.3184 \\ a_3 &= 745.0609 \end{aligned}$$

$$\frac{\sigma_{25}}{\sigma_{18}} = 1.162067$$

$$0.1 < p < 1.0$$

$$X = \sum_{n=0}^{n=4} a_n (\sqrt{p})^n \quad (4)$$

where

$$\begin{aligned} a_0 &= 53.6508 \\ a_1 &= 17.7272 \\ a_2 &= -6.9940 \\ a_3 &= -2.0216 \\ a_4 &= 3.0262 \end{aligned}$$

$$1.0 < p < 10.0$$

$$X = \sum_{n=0}^{n=3} a_n \ln(p)^n \quad (5)$$

where

$$\begin{aligned} a_0 &= 65.3068 \\ a_1 &= 7.0523 \\ a_2 &= -3.1346 \\ a_3 &= 1.4293 \end{aligned}$$

$$10 < p < 26$$

$$X = \sum_{n=0}^{n=4} a_n \ln(p)^n \quad (6)$$

where

$$\begin{aligned} a_0 &= -923.8866 \\ a_1 &= 1241.9749 \\ a_2 &= -546.7730 \\ a_3 &= 96.6712 \\ a_4 &= -4.8034 \end{aligned}$$

*Corrected by me*

Using the above equations a table of specific resistivity against concentration was prepared by means of the computer. This is presented here as table I. The accuracy of the data presented is about  $\pm 4\%$  after taking into account the accuracy of the original data.

### 3. Temperature coefficient

Coefficients ( $b_t$ ), as defined by equation 1, were calculated for all the available data. The coefficients were independent of concentration for concentration below 0.1% and increased continuously from 0.021 - 0.027 in the temperature range 0 - 140°C. The variation with temperature was fitted by the equation:

$$b_t = \sum_{n=0}^{n=4} a_n t^n \quad (7)$$

where

$$\begin{aligned} a_0 &= 2.1179818 \times 10^{-2} \\ a_1 &= 7.8601061 \times 10^{-5} \\ a_2 &= 1.5439826 \times 10^{-7} \\ a_3 &= -6.2634979 \times 10^{-9} \\ a_4 &= 2.2794885 \times 10^{-11} \end{aligned}$$

Values of  $b_t$  are tabulated in table II. The effect of concentrations on  $b_t$  is only slight at moderate concentrations. Tentative corrections for the effect of concentration on  $b_t$  are given in table III - insufficient data exists to give more accurate figures.

### 4. Summary of methods of use:

The use of the tables is best illustrated by taking two examples as follows:

(a) What is the specific resistivity of a sodium chloride solution containing 0.024% NaCl at 63.4°C?

Resistivity at 18°C (Table I) = 2342.6 ohm. cm.

Value of  $b_t$  at 63.4°C (Table II) = 0.02552

Concentration correction to  $b_t$  (Table III) = 0

Resistivity at 63.4°C (Equation 1)

$$= \frac{2342.6}{1 + 0.02552 (63.4 - 18)} = 1085.2 \text{ ohm. cm.}$$

(b) What is the concentration of a sodium chloride solution which has a specific resistivity of 110.4 ohm. cm. at 31.7°C?

$$b_t \text{ at } 31.7^\circ\text{C} = 0.02364$$

$$\begin{aligned} \text{Resistivity at } 18^\circ\text{C} &= 110.4 \times (1 + 0.02364 (31.7 - 18)) \\ &= 146.16 \text{ ohm. cm.} \end{aligned}$$

$$\text{Concentration (Table I)} = 0.425\%$$

$$\text{Concentration correction to } b_t \text{ (Table III)} = -0.0004$$

$$\text{New value of } b_t = 0.02324$$

$$\begin{aligned} \text{Resistivity at } 18^\circ\text{C} &= 110.4 \times (1 + 0.02324 (31.7 - 18)) \\ &= 145.56 \text{ ohm. cm.} \end{aligned}$$

$$\text{Concentration} = 0.428\%$$

The overall accuracy of calculations as described above is about  $\pm \frac{1}{2}\%$ .

#### Reference

- (1) Hewitt, G.F. Ph.D. Thesis, Manchester, 1957.

TABLE I.

## SODIUM CHLORIDE SOLUTION RESISTIVITIES AT 18°C. (OHM.CM.)

% NaCl	0	2	4	6	8	10
0.0010	54210	53152	52136	51157	50215	49306
0.0011	49306	48431	47586	46770	45981	45219
0.0012	45219	44482	43768	43077	42408	41759
0.0013	41759	41130	40520	39927	39352	38793
0.0014	38793	38250	37722	37208	36708	36222
0.0015	36222	35748	35287	34837	34399	33971
0.0016	33971	33554	33148	32751	32363	31985
0.0017	31985	31615	31254	30901	30556	30219
0.0018	30219	29889	29566	29250	28941	28639
0.0019	28639	28342	28052	27768	27489	27216
0.0020	27216	26948	26686	26428	26176	25928
0.0021	25928	25686	25447	25213	24983	24758
0.0022	24758	24536	24319	24105	23895	23689
0.0023	23689	23486	23287	23091	22898	22709
0.0024	22709	22522	22339	22159	21981	21807
0.0025	21807	21635	21466	21299	21136	20974
0.0026	20974	20815	20659	20505	20353	20203
0.0027	20203	20056	19910	19767	19626	19487
0.0028	19487	19350	19215	19081	18950	18820
0.0029	18820	18692	18566	18442	18319	18198
0.0030	18198	18078	17960	17844	17729	17615
0.0031	17615	17503	17393	17283	17176	17069
0.0032	17069	16964	16860	16757	16656	16556
0.0033	16556	16457	16359	16263	16167	16073
0.0034	16073	15980	15888	15797	15707	15618
0.0035	15618	15530	15443	15357	15272	15187
0.0036	15187	15104	15022	14941	14860	14780
0.0037	14780	14702	14624	14547	14470	14395
0.0038	14395	14320	14246	14173	14101	14029
0.0039	14029	13958	13888	13818	13750	13681
0.0040	13681	13614	13547	13481	13416	13351
0.0041	13351	13286	13223	13160	13098	13036
0.0042	13036	12975	12914	12854	12794	12735
0.0043	12735	12677	12619	12562	12505	12449
0.0044	12449	12393	12338	12283	12228	12175
0.0045	12175	12121	12068	12016	11964	11912
0.0046	11912	11861	11811	11761	11711	11661
0.0047	11661	11613	11564	11516	11468	11421
0.0048	11421	11374	11327	11281	11236	11190
0.0049	11190	11145	11100	11056	11012	10969
0.0050	10969	10925	10882	10840	10798	10756
0.0051	10756	10714	10673	10632	10591	10551
0.0052	10551	10511	10471	10432	10393	10354
0.0053	10354	10315	10277	10239	10202	10164
0.0054	10164	10127	10090	10054	10017	9981

TABLE I CONTINUED

%NaCl	0	2	4	6	8	10
0.0055	9981	9946	9910	9875	9840	9805
0.0056	9805	9771	9736	9702	9669	9635
0.0057	9635	9602	9569	9536	9503	9471
0.0058	9471	9438	9407	9375	9343	9312
0.0059	9312	9281	9250	9219	9189	9158
0.0060	9158	9128	9099	9069	9039	9010
0.0061	9010	8981	8952	8923	8895	8866
0.0062	8866	8838	8810	8782	8755	8727
0.0063	8727	8700	8673	8646	8619	8593
0.0064	8593	8566	8540	8514	8488	8462
0.0065	8462	8436	8411	8385	8360	8335
0.0066	8335	8310	8286	8261	8237	8212
0.0067	8212	8188	8164	8140	8117	8093
0.0068	8093	8070	8046	8023	8000	7977
0.0069	7977	7954	7932	7909	7887	7865
0.0070	7865	7843	7821	7799	7777	7755
0.0071	7755	7734	7712	7691	7670	7649
0.0072	7649	7628	7607	7587	7566	7546
0.0073	7546	7525	7505	7485	7465	7445
0.0074	7445	7425	7405	7386	7366	7347
0.0075	7347	7328	7309	7289	7270	7252
0.0076	7252	7233	7214	7196	7177	7159
0.0077	7159	7140	7122	7104	7086	7068
0.0078	7068	7050	7033	7015	6997	6980
0.0079	6980	6963	6945	6928	6911	6894
0.0080	6894	6877	6860	6843	6827	6810
0.0081	6810	6793	6777	6761	6744	6728
0.0082	6728	6712	6696	6680	6664	6648
0.0083	6648	6632	6617	6601	6586	6570
0.0084	6570	6555	6540	6524	6509	6494
0.0085	6494	6479	6464	6449	6434	6420
0.0086	6420	6405	6390	6376	6361	6347
0.0087	6347	6333	6318	6304	6290	6276
0.0088	6276	6262	6248	6234	6220	6206
0.0089	6206	6193	6179	6166	6152	6139
0.0090	6139	6125	6112	6099	6085	6072
0.0091	6072	6059	6046	6033	6020	6007
0.0092	6007	5994	5982	5969	5956	5944
0.0093	5944	5931	5919	5906	5894	5881
0.0094	5881	5869	5857	5845	5833	5821
0.0095	5821	5808	5797	5785	5773	5761
0.0096	5761	5749	5737	5726	5714	5702
0.0097	5702	5691	5679	5668	5657	5645
0.0098	5645	5634	5623	5611	5600	5589
0.0099	5589	5578	5567	5556	5545	5534

TABLE I CONTINUED

% NaCl	0	2	4	6	8	10
0.010	5534.2	5428.2	5325.4	5226.4	5131.1	5039.2
0.011	5039.2	4950.6	4865.1	4782.5	4702.7	4625.5
0.012	4625.5	4550.9	4478.7	4408.7	4341.0	4275.3
0.013	4275.3	4211.6	4149.7	4089.7	4031.5	3974.8
0.014	3974.8	3919.8	3866.3	3814.2	3763.6	3714.3
0.015	3714.3	3666.2	3619.5	3573.9	3529.4	3486.1
0.016	3486.1	3443.9	3402.6	3362.4	3323.1	3284.7
0.017	3284.7	3247.2	3210.6	3174.8	3139.7	3105.5
0.018	3105.5	3072.0	3039.3	3007.2	2975.8	2945.1
0.019	2945.1	2915.0	2885.6	2856.7	2828.4	2800.7
0.020	2800.7	2773.5	2746.8	2720.7	2695.1	2669.9
0.021	2669.9	2645.2	2621.0	2597.2	2573.9	2551.0
0.022	2551.0	2528.5	2506.3	2484.6	2463.3	2442.3
0.023	2442.3	2421.7	2401.4	2381.5	2361.9	2342.6
0.024	2342.6	2323.7	2305.1	2286.7	2268.7	2250.9
0.025	2250.9	2233.4	2216.2	2199.3	2182.6	2166.2
0.026	2166.2	2150.0	2134.1	2118.4	2102.9	2087.7
0.027	2087.7	2072.7	2057.9	2043.3	2029.0	2014.8
0.028	2014.8	2000.8	1987.1	1973.5	1960.1	1946.9
0.029	1946.9	1933.8	1921.0	1908.3	1895.8	1883.5
0.030	1883.5	1871.3	1859.2	1847.4	1835.7	1824.1
0.031	1824.1	1812.7	1801.4	1790.3	1779.3	1768.4
0.032	1768.4	1757.7	1747.1	1736.7	1726.3	1716.1
0.033	1716.1	1706.0	1696.0	1686.2	1676.5	1666.8
0.034	1666.8	1657.3	1647.9	1638.6	1629.5	1620.4
0.035	1620.4	1611.4	1602.5	1593.7	1585.0	1576.5
0.036	1576.5	1568.0	1559.6	1551.3	1543.0	1534.9
0.037	1534.9	1526.9	1518.9	1511.0	1503.2	1495.5
0.038	1495.5	1487.9	1480.3	1472.9	1465.5	1458.2
0.039	1458.2	1450.9	1443.7	1436.6	1429.6	1422.6
0.040	1422.6	1415.7	1408.9	1402.1	1395.5	1388.8
0.041	1388.8	1382.3	1375.8	1369.3	1362.9	1356.6
0.042	1356.6	1350.4	1344.3	1338.0	1331.9	1325.9
0.043	1325.9	1319.9	1314.0	1308.1	1302.3	1296.6
0.044	1296.6	1290.9	1285.3	1279.6	1274.0	1268.5
0.045	1268.5	1263.1	1257.6	1252.3	1247.0	1241.7
0.046	1241.7	1236.5	1231.3	1226.1	1221.0	1216.0
0.047	1216.0	1211.0	1206.0	1201.1	1196.3	1191.3
0.048	1191.3	1186.5	1181.8	1177.0	1172.4	1167.7
0.049	1167.7	1163.1	1158.5	1154.0	1149.5	1145.0
0.050	1145.0	1140.6	1136.2	1131.8	1127.5	1123.3
0.051	1123.3	1118.9	1114.7	1110.5	1106.3	1102.2
0.052	1102.2	1098.1	1094.0	1090.0	1086.0	1082.0
0.053	1082.0	1078.0	1074.1	1070.3	1066.4	1062.5
0.054	1062.5	1058.7	1054.9	1051.2	1047.5	1043.8

TABLE I CONTINUED

% NaCl	0	2	4	6	8	10
0.055	1043.8	1040.1	1036.4	1032.8	1029.2	1025.7
0.056	1025.7	1022.1	1018.6	1015.1	1011.6	1008.3
0.057	1008.2	1004.8	1001.4	998.0	994.6	991.3
0.058	991.3	988.0	984.7	981.5	978.3	975.0
0.059	975.0	971.8	968.6	965.5	962.4	959.2
0.060	959.2	956.3	953.1	950.0	947.0	944.0
0.061	944.0	941.0	938.0	935.1	932.1	929.2
0.062	929.2	926.3	923.4	920.6	917.7	914.9
0.063	914.9	912.1	909.3	906.5	903.8	901.1
0.064	901.1	898.3	895.6	892.9	890.3	887.6
0.065	887.6	885.0	882.4	879.7	877.3	874.6
0.066	874.6	872.0	869.5	866.9	864.4	861.9
0.067	861.9	859.4	857.0	854.5	852.1	849.6
0.068	849.6	847.2	844.8	842.4	840.1	837.7
0.069	837.7	835.4	833.0	830.7	828.4	826.1
0.070	826.1	823.8	821.6	819.3	817.1	814.9
0.071	814.9	812.6	810.4	808.2	806.1	803.9
0.072	803.9	801.7	799.6	797.5	795.3	793.2
0.073	793.2	791.1	789.0	787.0	784.9	782.9
0.074	782.9	780.8	778.8	776.8	774.8	772.8
0.075	772.8	770.8	768.8	766.8	764.9	762.9
0.076	762.9	761.0	759.0	757.1	755.2	753.3
0.077	753.3	751.4	749.6	747.7	745.8	744.0
0.078	744.0	742.1	740.3	738.5	736.7	734.9
0.079	734.9	733.1	731.3	729.5	727.7	726.0
0.080	726.0	724.2	722.5	720.8	719.0	717.3
0.081	717.3	715.6	713.9	712.2	710.5	708.9
0.082	708.9	707.2	705.5	703.9	702.2	700.6
0.083	700.6	699.0	697.4	695.7	694.1	692.5
0.084	692.5	690.9	689.4	687.8	686.2	684.7
0.085	684.7	683.1	681.6	680.0	678.5	677.0
0.086	677.0	675.5	673.9	672.4	670.9	669.4
0.087	669.4	668.0	666.5	665.0	663.6	662.1
0.088	662.1	660.6	659.2	657.8	656.3	654.9
0.089	654.9	653.5	652.1	650.7	649.3	647.9
0.090	647.9	646.5	645.1	643.7	642.4	641.0
0.091	641.0	639.6	638.3	637.0	635.6	634.3
0.092	634.3	632.9	631.6	630.3	629.0	627.7
0.093	627.7	626.4	625.1	623.8	622.5	621.2
0.094	621.2	620.0	618.7	617.4	616.2	614.9
0.095	614.9	613.7	612.4	611.2	610.0	608.7
0.096	608.7	607.5	606.3	605.1	603.9	602.7
0.097	602.7	601.5	600.3	599.1	597.9	596.8
0.098	596.8	595.6	594.4	593.2	592.1	590.9
0.099	590.9	589.8	588.6	587.5	586.4	585.2



TABLE I CONTINUED

% NaCl	0	2	4	6	8	10
0.10	585.24	574.16	563.51	553.26	543.38	533.85
0.11	533.85	524.66	515.79	507.22	498.94	490.93
0.12	490.93	483.18	475.68	468.41	461.37	454.54
0.13	454.54	447.91	441.48	435.23	429.17	423.27
0.14	423.27	417.54	411.96	406.54	401.26	396.12
0.15	396.12	391.11	386.23	381.48	376.84	372.32
0.16	372.32	367.90	363.60	359.39	355.29	351.27
0.17	351.27	347.36	343.52	339.78	336.12	332.54
0.18	332.54	329.03	325.60	322.25	318.96	315.74
0.19	315.74	312.59	309.50	306.47	303.51	300.60
0.20	300.60	297.75	294.95	292.21	289.52	286.88
0.21	286.88	284.29	281.74	279.25	276.79	274.39
0.22	274.39	272.02	269.69	267.41	265.17	262.96
0.23	262.96	260.79	258.66	256.56	254.50	252.47
0.24	252.47	250.48	248.51	246.58	244.68	242.81
0.25	242.81	240.97	239.15	237.37	235.61	233.88
0.26	233.88	232.17	230.49	228.83	227.20	225.60
0.27	225.60	224.01	222.45	220.91	219.39	217.89
0.28	217.89	216.42	214.96	213.53	212.11	210.72
0.29	210.72	209.34	207.98	206.64	205.31	204.01
0.30	204.01	202.72	201.45	200.19	198.95	197.72
0.31	197.72	196.52	195.32	194.14	192.98	191.83
0.32	191.83	190.69	189.57	188.46	187.36	186.28
0.33	186.28	185.21	184.15	183.11	182.08	181.05
0.34	181.05	180.05	179.05	178.06	177.09	176.12
0.35	176.12	175.17	174.22	173.29	172.37	171.46
0.36	171.46	170.56	169.66	168.78	167.91	167.04
0.37	167.04	166.19	165.34	164.50	163.67	162.85
0.38	162.85	162.04	161.24	160.44	159.66	158.88
0.39	158.88	158.10	157.34	156.58	155.84	155.09
0.40	155.09	154.36	153.63	152.91	152.20	151.49
0.41	151.49	150.79	150.10	149.41	148.73	148.06
0.42	148.06	147.39	146.73	146.08	145.43	144.78
0.43	144.78	144.15	143.52	142.89	142.27	141.65
0.44	141.65	141.05	140.44	139.84	139.25	138.66
0.45	138.66	138.08	137.50	136.93	136.36	135.80
0.46	135.80	135.24	134.68	134.14	133.59	133.05
0.47	133.05	132.52	131.98	131.46	130.94	130.42
0.48	130.42	129.90	129.39	128.89	128.39	127.89
0.49	127.89	127.40	126.91	126.42	125.94	125.46
0.50	125.46	124.99	124.52	124.05	123.59	123.13
0.51	123.13	122.67	122.22	121.77	121.32	120.88
0.52	120.88	120.44	120.00	119.57	119.14	118.72
0.53	118.72	118.29	117.87	117.46	117.04	116.63
0.54	116.63	116.22	115.82	115.42	115.02	114.62

TABLE I CONTINUED

% NaCl	0	2	4	6	8	10
0.55	114.62	114.23	113.84	113.45	113.06	112.68
0.56	112.68	112.30	111.92	111.55	111.18	110.81
0.57	110.81	110.44	110.08	109.72	109.36	109.00
0.58	109.00	108.65	108.29	107.94	107.60	107.25
0.59	107.25	106.91	106.57	106.23	105.89	105.56
0.60	105.56	105.23	104.90	104.57	104.25	103.92
0.61	103.92	103.60	103.28	102.97	102.65	102.34
0.62	102.34	102.03	101.72	101.41	101.11	100.80
0.63	100.80	100.50	100.20	99.90	99.61	99.31
0.64	99.31	99.02	98.73	98.44	98.16	97.87
0.65	97.87	97.59	97.31	97.03	96.75	96.47
0.66	96.47	96.20	95.92	95.65	95.38	95.11
0.67	95.11	94.84	94.58	94.32	94.05	93.79
0.68	93.79	93.53	93.28	93.02	92.76	92.51
0.69	92.51	92.26	92.01	91.76	91.51	91.26
0.70	91.26	91.02	90.78	90.53	90.29	90.05
0.71	90.05	89.82	89.58	89.34	89.11	88.88
0.72	88.88	88.64	88.41	88.18	87.96	87.73
0.73	87.73	87.50	87.28	87.06	86.84	86.61
0.74	86.61	86.40	86.18	85.96	85.74	85.53
0.75	85.53	85.32	85.10	84.89	84.68	84.47
0.76	84.47	84.26	84.06	83.85	83.64	83.44
0.77	83.44	83.24	83.04	82.83	82.63	82.44
0.78	82.44	82.24	82.04	81.85	81.65	81.46
0.79	81.46	81.26	81.07	80.88	80.69	80.50
0.80	80.50	80.31	80.13	79.94	79.75	79.57
0.81	79.57	79.39	79.20	79.02	78.84	78.66
0.82	78.66	78.48	78.30	78.13	77.95	77.77
0.83	77.77	77.60	77.42	77.25	77.08	76.91
0.84	76.91	76.74	76.57	76.40	76.23	76.06
0.85	76.06	75.89	75.73	75.56	75.40	75.23
0.86	75.23	75.07	74.91	74.75	74.59	74.43
0.87	74.43	74.27	74.11	73.95	73.79	73.64
0.88	73.64	73.48	73.33	73.17	73.02	72.86
0.89	72.86	72.71	72.56	72.41	72.26	72.11
0.90	72.11	71.96	71.81	71.66	71.52	71.37
0.91	71.37	71.23	71.08	70.94	70.79	70.65
0.92	70.65	70.51	70.36	70.22	70.08	69.94
0.93	69.94	69.80	69.66	69.53	69.39	69.25
0.94	69.25	69.11	68.98	68.84	68.71	68.57
0.95	68.57	68.44	68.31	68.17	68.04	67.91
0.96	67.91	67.78	67.65	67.52	67.39	67.26
0.97	67.26	67.13	67.00	66.88	66.75	66.62
0.98	66.62	66.50	66.37	66.25	66.12	66.00
0.99	66.00	65.88	65.75	65.62	65.50	65.38

TABLE I CONTINUED

% NaCl	0	2	4	6	8	10
1.0	65.387	64.162	63.056	61.988	60.955	59.956
1.1	59.956	58.989	58.053	57.146	56.267	55.414
1.2	55.414	54.587	53.785	53.005	52.249	51.513
1.3	51.513	50.798	50.103	49.427	48.769	48.128
1.4	48.128	47.504	46.896	46.304	45.727	45.164
1.5	45.164	44.615	44.080	43.557	43.047	42.548
1.6	42.548	42.062	41.586	41.122	40.667	40.223
1.7	40.223	39.789	39.364	38.949	38.542	38.144
1.8	38.144	37.754	37.373	36.999	36.633	36.274
1.9	36.274	35.922	35.577	35.239	34.908	34.583
2.0	34.583	34.264	33.951	33.644	33.342	33.046
2.1	33.046	32.756	32.471	32.191	31.915	31.645
2.2	31.645	31.379	31.118	30.862	30.610	30.362
2.3	30.362	30.118	29.878	29.642	29.410	29.182
2.4	29.182	28.958	28.737	28.519	28.305	28.095
2.5	28.095	27.887	27.683	27.482	27.284	27.089
2.6	27.089	26.896	26.707	26.520	26.337	26.155
2.7	26.155	25.977	25.801	25.627	25.456	25.288
2.8	25.288	25.121	24.957	24.795	24.636	24.478
2.9	24.478	24.323	24.170	24.019	23.869	23.722
3.0	23.722	23.577	23.433	23.292	23.152	23.014
3.1	23.014	22.878	22.743	22.610	22.479	22.349
3.2	22.349	22.221	22.095	21.970	21.846	21.725
3.3	21.725	21.604	21.485	21.367	21.251	21.136
3.4	21.136	21.022	20.910	20.799	20.689	20.581
3.5	20.581	20.473	20.367	20.262	20.159	20.056
3.6	20.056	19.954	19.854	19.755	19.656	19.559
3.7	19.559	19.463	19.368	19.274	19.181	19.088
3.8	19.088	18.997	18.907	18.817	18.729	18.641
3.9	18.641	18.555	18.469	18.384	18.300	18.217
4.0	18.217	18.134	18.053	17.972	17.892	17.813
4.1	17.813	17.734	17.656	17.579	17.503	17.428
4.2	17.428	17.353	17.279	17.205	17.133	17.060
4.3	17.060	16.989	16.918	16.848	16.779	16.710
4.4	16.710	16.642	16.574	16.507	16.441	16.375
4.5	16.375	16.309	16.245	16.181	16.117	16.054
4.6	16.054	15.992	15.930	15.868	15.807	15.747
4.7	15.747	15.687	15.628	15.569	15.511	15.453
4.8	15.453	15.395	15.338	15.282	15.226	15.170
4.9	15.170	15.115	15.061	15.006	14.953	14.899
5.0	14.899	14.846	14.794	14.742	14.690	14.639
5.1	14.639	14.588	14.537	14.487	14.438	14.388
5.2	14.388	14.339	14.291	14.242	14.195	14.147
5.3	14.147	14.100	14.053	14.007	13.961	13.915
5.4	13.915	13.869	13.824	13.780	13.735	13.691

TABLE I CONTINUED

% NaCl	0	2	4	6	8	10
5.5	13.691	13.647	13.604	13.561	13.518	13.475
5.6	13.475	13.433	13.391	13.349	13.308	13.267
5.7	13.267	13.226	13.186	13.145	13.105	13.066
5.8	13.066	13.026	12.987	12.948	12.910	12.871
5.9	12.871	12.833	12.796	12.758	12.721	12.684
6.0	12.684	12.647	12.610	12.574	12.538	12.502
6.1	12.502	12.466	12.431	12.396	12.361	12.326
6.2	12.326	12.291	12.257	12.223	12.189	12.156
6.3	12.156	12.122	12.089	12.056	12.023	11.990
6.4	11.990	11.958	11.926	11.894	11.862	11.831
6.5	11.831	11.799	11.768	11.737	11.706	11.675
6.6	11.675	11.645	11.615	11.584	11.555	11.525
6.7	11.525	11.495	11.466	11.437	11.408	11.379
6.8	11.379	11.350	11.321	11.293	11.265	11.237
6.9	11.237	11.209	11.181	11.154	11.126	11.099
7.0	11.099	11.072	11.045	11.018	10.991	10.965
7.1	10.965	10.938	10.912	10.886	10.860	10.835
7.2	10.835	10.809	10.783	10.758	10.733	10.708
7.3	10.708	10.683	10.658	10.633	10.609	10.584
7.4	10.584	10.560	10.536	10.512	10.488	10.464
7.5	10.464	10.441	10.417	10.394	10.371	10.347
7.6	10.347	10.324	10.301	10.279	10.256	10.233
7.7	10.233	10.211	10.189	10.166	10.144	10.122
7.8	10.122	10.101	10.079	10.057	10.036	10.014
7.9	10.014	9.993	9.972	9.951	9.930	9.909
8.0	9.909	9.888	9.867	9.847	9.826	9.806
8.1	9.806	9.785	9.765	9.745	9.725	9.705
8.2	9.705	9.685	9.666	9.646	9.627	9.607
8.3	9.607	9.588	9.569	9.549	9.530	9.511
8.4	9.511	9.493	9.474	9.455	9.436	9.418
8.5	9.418	9.399	9.381	9.363	9.345	9.327
8.6	9.327	9.308	9.291	9.273	9.255	9.237
8.7	9.237	9.220	9.202	9.185	9.167	9.150
8.8	9.150	9.133	9.116	9.099	9.082	9.065
8.9	9.065	9.048	9.031	9.014	8.998	8.981
9.0	8.981	8.965	8.948	8.932	8.916	8.900
9.1	8.900	8.883	8.867	8.851	8.836	8.820
9.2	8.820	8.804	8.788	8.773	8.757	8.741
9.3	8.741	8.726	8.711	8.695	8.680	8.665
9.4	8.665	8.650	8.635	8.620	8.605	8.590
9.5	8.590	8.575	8.560	8.546	8.531	8.517
9.6	8.517	8.502	8.488	8.473	8.459	8.445
9.7	8.445	8.430	8.416	8.402	8.388	8.374
9.8	8.374	8.360	8.346	8.333	8.319	8.305
9.9	8.305	8.292	8.278	8.264	8.251	8.237

TABLE I CONTINUED

% NaCl	0	2	4	8	8	10
I0	8.237	8.123	8.010	7.898	7.787	7.679
I1	7.674	7.572	7.473	7.376	7.281	7.189
I2	7.189	7.099	7.011	6.926	6.843	6.762
I3	6.762	6.684	6.608	6.535	6.463	6.394
I4	6.394	6.327	6.262	6.199	6.138	6.079
I5	6.079	6.022	5.966	5.913	5.861	5.811
I6	5.811	5.763	5.716	5.670	5.627	5.584
I7	5.584	5.543	5.504	5.465	5.428	5.393
I8	5.393	5.358	5.325	5.292	5.261	5.231
I9	5.231	5.202	5.174	5.147	5.120	5.095
20	5.095	5.071	5.047	5.024	5.002	4.981
21	4.981	4.961	4.941	4.922	4.903	4.886
22	4.886	4.868	4.852	4.836	4.821	4.806
23	4.806	4.792	4.778	4.765	4.752	4.740
24	4.740	4.728	4.716	4.705	4.695	4.685
25	4.685	4.675	4.666	4.657	4.648	4.640
26	4.640	4.632	4.624			

TABLE 2 Values of  $\alpha_T$

TEMPERATURE COEFFICIENTS FOR LOW-CONCENTRATION SODIUM CHLORIDE SOLUTIONS

TEMP °C.	0	2	4	6	8	10
0	0.02118	0.02134	0.02150	0.02166	0.02182	0.02198
10	0.02198	0.02213	0.02229	0.02245	0.02261	0.02277
20	0.02277	0.02292	0.02308	0.02323	0.02338	0.02353
30	0.02353	0.02367	0.02382	0.02396	0.02409	0.02423
40	0.02423	0.02436	0.02449	0.02461	0.02474	0.02486
50	0.02486	0.02497	0.02508	0.02519	0.02529	0.02539
60	0.02539	0.02549	0.02558	0.02567	0.02576	0.02584
70	0.02584	0.02591	0.02599	0.02606	0.02612	0.02618
80	0.02618	0.02624	0.02629	0.02634	0.02639	0.02643
90	0.02643	0.02647	0.02651	0.02654	0.02657	0.02660
100	0.02660	0.02662	0.02665	0.02666	0.02668	0.02669
110	0.02669	0.02671	0.02672	0.02673	0.02673	0.02674
120	0.02674	0.02674	0.02675	0.02675	0.02675	0.02676
130	0.02676	0.02676	0.02676	0.02677	0.02677	0.02678
140	0.02678	0.02679	0.02680	0.02681	0.02683	0.02684

TABLE 3

TENTATIVE CORRECTIONS TO SODIUM CHLORIDE SOLUTION TEMPERATURE COEFFICIENTS

TEMP °C.	% NaCl			
	0.1	0.5	1.0	5.0
0	-0.0001	-0.0004	-0.0009	-0.0011
50	-0.0001	-0.0007	-0.0007	-0.0004
100	-0.0002	-0.0012	-0.0004	+0.0004